

The Structure of *N*-Methyl-*p*-chlorobenzaldoxime and Refinement of the Structures of 'syn'- and 'anti'-*p*-Chlorobenzaldoxime

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N-Methyl-*p*-chlorobenzaldoxime, prepared from 'syn'-*p*-chlorobenzaldoxime by use of dimethyl sulfate, is shown by a three-dimensional X-ray diffraction study to have the *anti* configuration. There are four molecules of $\text{ClC}_6\text{H}_4\text{CHN}(\text{CH}_3)\text{O}$ in a unit cell having parameters $a = 7.50$, $b = 9.91$, $c = 11.68 \text{ \AA}$ and $\beta = 108^\circ 1'$. The space group is $P2_1/a$. For the 936 observed diffraction maxima the value of $R = \sum ||F_o| - |F_c|| / \sum |F_o|$ is 0.085.

Refinement of the previously determined structure of 'syn'-*p*-chlorobenzaldoxime by three-dimensional least-squares methods has yielded $R = 0.116$ for the 1238 observed maxima. A refinement of two-dimensional data for 'anti'-*p*-chlorobenzaldoxime has also been made, but better data are needed for an accurate structure. A comparison of $\text{C}=\text{N}$ and $\text{N}-\text{O}$ distances in this $\text{C}=\text{N}-\text{O}$ system is presented.

This study of the crystal and molecular structure of *N*-methyl-*p*-chlorobenzaldoxime and refinements of the structures of 'syn'- and 'anti'-*p*-chlorobenzaldoxime form part of a series of studies of the $\text{C}=\text{N}-\text{O}$ bond system in oximes $\text{RR}'\text{C}=\text{NOH}$ and their alkylated or acylated derivatives. It is expected that detailed information on bond angles and distances will elucidate the nature of bonding in oximes and amine oxides, the questions of relative stabilities of 'syn' and 'anti' isomers, the relation of structure to *N*- versus *O*-alkylation, and problems relating to planarity, conjugation, charge transfer and spectra of these oximes and their derivatives.

The configuration of the *N*-methyl derivative of oximes has been studied earlier by chemical methods. Methylation by use of dimethyl sulfate, which has never yielded both of the theoretically possible *N*-methyl isomers, has been said to yield the *anti* configuration (Brady & Dunn, 1926; Sutton & Taylor, 1931). On the other hand, use of diazomethane as a methylating agent is stated to yield this same *N*-methyl isomer from the 'syn' oxime, but a different *N*-methyl isomer from the 'anti' oxime (Thompson & Baer, 1940). Our study shows conclusively that dimethyl sulfate yields the 'anti' configuration for the *N*-methyl derivative obtained from 'syn'-*p*-chlorobenzaldoxime.

The relation of this new structure to the known structures of oximes and amine oxides is exhibited below, but a comparison of *O*-methyl and *N*-methyl derivatives, both of which are known (Brady, Dunn & Goldstein, 1926), will have to await a projected crystal structure study of an *O*-methyl oxime.

In addition, we describe least-squares refinements

of the X-ray data which were used previously (Jerslev, 1958) in the elucidation of the structures of 'syn'- and 'anti'-*p*-chlorobenzaldoximes, which are closely related to the new structure of *N*-methyl-*p*-chlorobenzaldoxime. The three-dimensional data of the 'syn'-*p*-chlorobenzaldoxime have yielded an accurate structure. However, as already pointed out (Jerslev, 1958), the slight inconsistencies in the data probably arising from cutting of the crystals of 'anti'-*p*-chlorobenzaldoxime to a suitable shape for X-ray study have limited the accuracy of refinement of the two-dimensional data for this substance. Nevertheless, these new refinements are presented for comparison of the results of crystallographic studies on these three compounds, and for addition to the review of data available on these and related compounds (Jerslev, 1958).

Experimental study and X-ray data

N-Methyl-*p*-chlorobenzaldoxime was prepared by reacting *p*-chlorobenzaldehyde with hydroxylamine hydrochloride (Erdmann & Schwechten, 1890) and methylating the resulting 'syn'-*p*-chlorobenzaldoxime (Brady, Dunn & Goldstein, 1926). Single crystals suitable for X-ray diffraction work were obtained by slow evaporation of a saturated solution in benzene. The melting point was 128–129.5°. Microanalysis confirmed the formula $\text{ClC}_6\text{H}_4\text{NO}$ (Cl: calc. 20.9, found 21.1; C: calc. 56.7, found 56.8; H: calc. 4.8, found 4.9; N: calc. 8.3, found 8.2%).

Single crystals no larger than 0.07 mm in cross section were chosen for study ($\mu = 35.95 \text{ cm}^{-1}$ for $\text{Cu } K\alpha$). From oscillation and Weissenberg photographs the reciprocal lattice symmetry was established

Table 1. List of observed and calculated structure factors for *N*-methyl-*p*-chlorobenzaldoxime

The three columns of each group contain from left to right the values of *l*, 10 F_o , and 10 F_c . The values of *l* followed by an asterisk indicate reflections below the observable limit. The value of F_o in this case is the minimum observed value. After this study was completed we found that the 185 reflection was indeed present on the films

	0, 0, L	0, 8, L	1, 3, L	1, 8, L	2, 2, L	2, 6, L
1	113 -110	0 139 147	4 367 -371	2* 22 -1	-12 59 55	-2 30 -28
2	402 392	1 85 96	5 68 -70	3 32 35	-11 39 -35	-1 102 -97
3	123 155	2 143 165	6 177 -177	4 31 29	-10 52 -51	0 82 68
4	147 122	3 67 -68	7 45 -36	5 33 -29	-8 50 -50	1 85 82
5	191 -180	4 32 35	8 195 -207	6* 21 10	-7 176 177	3 75 -77
6	90 76	5* 22 8	9 54 -53	7 29 -17	-6 48 -43	4 139 136
7	28 35	6 70 83	10 41 44	8 31 -27	-5 318 305	5 115 -116
8	208 -219	7 33 34	11 30 34	9* 16 -10	-4 113 -97	6* 24 -12
9	127 126	8 20 22	12 40 41	10 20 -23	-3 89 -89	7 90 89
10	184 -215	9* 18 -4	13 30 34	-	-1 52 -589	8 50 38
11	35 -11	10 49 -52	-	-	-1 422 450	9 43 35
12	14 -107	-	-	-	0 251 270	10 49 43
13	60 -68	-	-	-	1 206 -206	-
14	63 -77	-	-	-	2 199 198	-
		0, 9, L	1, 4, L	2, 3, L	2, 7, L	
0, 1, L	1 104 -123	-12 16 42	-8 65 -65	-1 206 -206	77 66	
1	2* 22 -9	-10 38 -41	-7* 20 -12	2 199 198	-8* 23 24	
2	3 28 -29	-7 43 48	-5* 21 11	4 80 77	-7 30 32	
3	4 24 -22	-6 132 -118	-3 96 108	5* 22 17	-6* 24 -12	
4	5 25 -25	-5 133 -118	-2 32 29	7 148 -149	-5 82 78	
5	6 25 -13	-4 66 -66	1 30 29	8 71 64	-4 167 -157	
6	560 -559	7 85 -82	0 1 1	9* 24 5	-5 82 78	
7	560 -559	8* 17 -18	-2 131 -126	10 34 -33	-4 167 -157	
8	201 -211	9 20 -19	-1 21 -6	2* 22 22	-3* 24 23	
9	93 -93	0, 10, L	0 270 -290	3* 22 -8	-1 65 60	
10	167 -177	1 188 95	5* 21 -7	-12 54 -44	76 -81	
11	514 -518	2 123 151	6 73 68	-11 32 -27	1 32 35	
12	26 -26	3 186 -191	7 32 -29	-10 26 -23	2 36 24	
13	34 -31	-	-	-9 103 -100	3 39 -44	
0, 2, L	0* 21 22	2 4 7	1, 10, L	-8 95 -81	37 22	
1	1 14 -117	5 70 61	-	80 -80	-	
2	2 114 -117	6 95 -87	-	-	-	
3	3* 21 2	7* 21 23	1, 1, L	185 -177	-	
4	4 50 -54	8 145 141	-9 13 -19	-5 185 -177	-	
5	0 142 150	5 36 -34	-8 36 -40	-5 175 130	-	
6	1 83 80	6* 18 20	7 35 -34	-4 94 90	7 47 49	
7	2 154 -153	7* 16 -1	11 36 38	-6 31 32	9 53 -10	
8	3 57 55	8 18 18	-	-2 116 107	10 27 27	
9	124 -119	-	-	209 193	-	
10	567 -160	0, 11, L	-	-	2, 8, L	
11	43 -32	-	-	-	-	
12	39 -37	1* 14 6	-11 60 52	-2 40 46	253 243	
13	564 -159	2 22 24	-10 38 38	1* 21 23	59 49	
14	69 -68	0, 12, L	-8 108 112	0 29 29	98 98	
15	25 -19	-	-7 35 34	1 127 129	-7 57 56	
16	26 30	0* 15 -22	-6 124 113	2 74 -73	-6* 24 23	
17	1 21 25	1 24 -9	-5* 19 17	3* 20 -4	-5 24 17	
18	2 23 -27	-3 33 -22	4* 19 -1	-7 54 -48	-4* 24 8	
19	3 26 23	-2 160 170	5* 18 -22	8 43 -42	-3 203 -217	
20	1 31 -23	-1 31 -23	6* 16 -13	9 83 -70	-2 77 -80	
21	1 210 -222	0 238 -273	7 36 -35	10 29 -26	-1* 23 -8	
22	1 151 -144	1 137 -137	-	11 67 -60	-1* 0 79 -83	
23	1 176 -188	1 137 -137	-	12 41 -40	1 31 77	
24	2 22 -25	-9 164 -163	2 18 -6	-	2 38 -27	
25	2 196 -199	3 33 24	-4 37 41	3 92 96		
26	2 231 -225	4 92 -87	-3* 18 -11	4* 24 12		
27	2 165 -145	5* 20 56	-2 32 33	-5 24 9		
28	2 110 -101	6 110 -101	-1* 19 -17	-6* 22 0		
29	3 118 -112	0 37 -43	-8 153 141	7 51 49		
30	3 151 -145	4 51 -39	1 51 -45	-7 80 69	8 36 33	
31	3 24 -20	5 24 -20	2* 18 -15	-6 94 93	9 32 35	
32	3 264 -277	6 25 26	3 45 -45	-5 173 148	-	
33	1 333 -311	7 10* 20 16	4 28 -27	290 270	2, 9, L	
34	0 420 -430	11 29 30	-	-	-	
35	1 27 -34	12 34 34	-	-	-	
0, 4, L	2 317 270	-	1, 12, L	-2 25 25	-8 69 -65	
1	3 205 242	-	-	75 68	-	
2	4 188 167	1, 6, L	-1 35 -25	0 250 250	3 92 96	
3	5 405 383	-12 20 22	0 22 -24	1 180 171	4* 24 12	
4	6 254 239	-11 49 45	1 23 -12	2 116 -105	-5 82 89	
5	6 58 54	-10* 20 14	-	3 98 -87	-3 42 -40	
6	5 43 44	-9 36 34	-	4 187 -178	-2 49 32	
7	0 45 44	-8 48 -50	-	5 245 -237	-1 118 124	
8	9* 23 3	-7 34 37	-13 60 53	6 239 -238	0 25 -29	
9	10* 22 -9	-6 40 39	-12 76 66	7 69 -63	1 22 84	
10	11 25 -23	-5 69 77	-11 169 -195	8 69 -59	2 83 73	
11	12 50 -38	-4 58 -56	-10 41 -32	9 69 -59	-3* 22 16	
12	13 36 -34	-3 184 -11	-9 44 -49	10 59 -55	4* 22 22	
13	42 43	-2 283 -283	-8 33 -27	11 51 -48	5 26 39	
14	49 58	1, 2, L	-7 237 -232	12 28 -28	31 39 33	
0, 5, L	-13 42 -54	0 95 -100	-5 68 62	-	-	
1	-11 71 -81	1 172 -165	-5 477 -464	2, 5, L	-8 69 -65	
2	-10 100 112	3 40 -42	-3 74 -11	2, 4, L	2 38 -27	
3	-9 98 -96	4 69 68	-2 1282 -1307	-7* 21 -14	-	
4	5 405 383	-12 20 22	0 23 -12	-11 60 60	-	
5	6 254 239	-11 49 45	-	5 127 132	-9* 24 27	
6	7* 22 0	-	2 87 -85	8* 23 -6	-7* 40 38	
7	8 18 15	8 38 42	3 808 778	-7 127 129	-6 40 32	
8	9 44 44	-7 78 73	4 38 -39	-6 26 -21	-5* 21 16	
9	10 39 38	10 67 67	5 209 212	-5 74 -74	-4 44 34	
10	11 51 -14	11 23 29	6 166 146	-4 122 -117	-3* 22 16	
11	12 60 -68	12 22 31	7 168 17	-3* 17 17	-3* 22 22	
12	13 61 -66	2 118 18	8 252 262	-2* 11 11	-1 46 -55	
13	1 421 448	-	-	-	0 47 48	
14	2 197 169	-8 126 -150	10 94 88	1 318 -326	1* 22 -7	
15	3 198 8	-7 20 -177	11 42 43	2 56 -49	2 50 48	
16	4 202 28	-6 97 -111	2 42 -37	3 37 -37	2* 19 2	
17	5 152 -143	-5 23 -26	-	5 199 -191	6 39 -46	
18	7 253 -246	-4 58 -58	-	6 27 -31	6 37 -26	
19	8 28 -23	-3 130 128	-12 47 -43	7* 24 -5	-	
20	9 150 -160	-2* 20 -1	-11* 24 -12	8* 23 -27	-	
21	10 48 -54	0 39 -37	-10 171 -185	9 14 -12	2, 11, L	
22	11 99 -96	0 83 87	-9 67 -67	10 34 -28	-	
23	12 60 -68	1 50 55	-8 97 -94	11 47 48	-1 57 48	
24	13 61 -66	2 118 18	-7 120 -118	0 64 -53	1 29 -23	
25	1 421 448	-	-	-	1 22 -25	
26	2 197 169	-8 126 -150	10 94 88	1 318 -326	1* 22 -7	
27	3 198 8	-7 20 -177	11 42 43	2 56 -49	2 50 48	
28	4 202 28	-6 97 -111	2 42 -37	3 37 -37	2* 19 2	
29	5 152 -143	-5 23 -26	-	5 199 -191	6 39 -46	
30	7 253 -246	-4 58 -58	-	6 27 -31	7 37 -26	
31	8 28 -23	-3 130 128	-12 47 -43	7* 24 -5	-	
32	9 150 -160	-2* 20 -1	-11* 24 -12	8* 23 -27	-	
33	10 48 -54	0 39 -37	-10 171 -185	9 14 -12	2, 11, L	
34	11 99 -96	0 83 87	-9 67 -67	10 34 -28	-	
35	12 60 -68	1 50 55	-8 97 -94	11 47 48	-1 57 48	
36	13 61 -66	2 118 18	-7 120 -118	0 64 -53	1 29 -23	
37	1 421 448	-	-	-	1 22 -25	
38	2 197 169	-8 126 -150	10 94 88	1 318 -326	1* 22 -7	
39	3 198 8	-7 20 -177	11 42 43	2 56 -49	2 50 48	
40	4 202 28	-6 97 -111	2 42 -37	3 37 -37	2* 19 2	
41	5 152 -143	-5 23 -26	-	5 199 -191	6 39 -46	
42	7 253 -246	-4 58 -58	-	6 27 -31	7 37 -26	
43	8 28 -23	-3 130 128	-12 47 -43	7* 24 -5	-	
44	9 150 -160	-2* 20 -1	-11* 24 -12	8* 23 -27	-	
45	10 48 -54	0 39 -37	-10 171 -185	9 14 -12	2, 11, L	
46	11 99 -96	0 83 87	-9 67 -67	10 34 -28	-	
47	12 60 -68	1 50 55	-8 97 -94	11 47 48	-1 57 48	
48	13 61 -66	2 118 18	-7 120 -118	0 64 -53	1 29 -23	
49	1 421 448	-	-	-	1 22 -25	
50	2 197 169	-8 126 -150	10 94 88	1 318 -326	1* 22 -7	
51	3 198 8	-7 20 -177	11 42 43	2 56 -49	2 50 48	
52	4 202 28	-6 97 -111	2 42 -37	3 37 -37	2* 19 2	
53	5 152 -143	-5 23 -26	-	5 199 -191	6 39 -46	
54	7 253 -246	-4 58 -58	-	6 27 -31	7 37 -26	
55	8 28 -23	-3 130 128	-12 47 -43	7* 24 -5	-	
56	9 150 -160	-2* 20 -1	-11* 24 -12	8* 23 -27	-	
57	10 48 -54	0 39 -37	-10 171 -185	9 14 -12	2, 11, L	
58	11 99 -96	0 83 87	-9 67 -67	10 34 -28	-	
59	12 60 -68	1 50 55	-8 97 -94	11 47 48	-1 57 48	
60	13 61 -66	2 118 18	-7 120 -118	0 64 -53	1 29 -23	
61	1 421 448	-	-	-	1 22 -25	
62	2 197 169	-8 126 -150	10 94 88	1 318 -326	1* 22 -7	
63	3 198 8	-7 20 -177	11 42 43	2 56 -49	2 50 48	
64	4 202 28	-6 97 -111	2 42 -37	3 37 -37	2* 19 2	
65	5 152 -143	-5 23 -26	-	5 199 -191	6 39 -46	
66	7 253 -246	-4 58 -58	-	6 27 -31	7 37 -26	
67	8 28 -23	-3 130 128	-12 47 -43	7* 24 -5	-	
68	9 150 -160	-2* 20 -1	-11* 24 -12	8* 23 -27	-	
69	10 48 -54	0 39 -37	-10 171 -185	9 14 -12	2, 11, L	
70	11 99 -96	0 83 87	-9 67 -67	10 34 -28	-	
71	12 60 -68	1 50 55	-8			

Table 1 (*cont.*)

3, 1, L	3, 6, L	4, 0, L	4, 6, L	5, 2, L	6, 1, L
-11 106 114	-10 96 89	5 93 -122	-10 61 -56	3 140 -154	0* -22 -78 -46
-10 98 88	-10 100 19	5 147 -148	-10 47 37	106 -105	1 -66 -23 -17
-9 61 54	-7* 28 19	7 42 -46	-9* 30 39	203 -205	2* 23 22 -17
-8 26 15	-6 58 56	8 46 39	-7* 54 -37	3 79 -80	3* 22 -22 -70
-7 96 93	-5 62 -43	4, 1, L	-6* 53 52	4 44 -27	4 65 -65 -70
-6 141 -134	-4 183 185	-11 44 31	-5* 95 87	5 46 -41	
-5 97 -85	-3 138 138	-11 51 41	-4* 29 7	6* 34 -8	
-4 316 -23	-2 39 -39	-10 38 -26	-2* 29 35	7 39 35	
-3 212 145	-1* 24 -1	-10 136 -135	-1 136 -135	5, 3, L	
-2 121 -201	1 31 44	-9 46 35	-2* 46 35	-1 49 58	-1 50 -55
-1 213 -166	2 139 -140	-8 58 25	0 46 35	1 53 -56	1 53 -61
0 167 -159	3 75 -78	-7 166 -17	1 46 -11	-7 54 -55	2* 27 -13
1 28 12	4 172 -176	-5* 58 -10	2 40 -35	-6* 27 -13	3 30 -54
2 300 -289	5 81 -90	-4 216 -223	3 40 24	-5 107 -110	
3 90 -86	6 74 -74	-3 179 -195	4 35 18	-4 154 -163	
4 85 -65	7* 26 -19	-2 191 -208	5 53 -59	-3 92 -91	
5* 24 6	8 -29 -36	-1 38 41	6 53 -60	-2 35 2	-8 54 -11
6* 66 6	0 158 -167	-1 160 156	7* 26 1	-1 30 -19	-6 173 188
7 25 18	1 160 156	8 43 -38	8 43 -38	0 42 27	-5* 32 -67
8 84 69	3, 7, L	1 138 -138	1 99 102	1 42 32	-3* 32 -45
9 73 66	2 39 33	2* 24 -4	2 56 -42	1 41 32	-3* 32 -37
10 75 67	9 47 40	4 31 -17	4 41 38	1 73 77	
11 3 36 32	8 36 32	5 56 70	5 81 94	0 73 77	
3, 2, L	7 37 -1	6 84 81	6 42 49	1 124 124	
-12 77 82	-6 51 54	7* 30 35	7 68 59	-5 113 113	
-11 46 -37	-5 189 -188	8 99 84	8 48 47	-4 61 63	-6 92 76
-10 62 -60	-4* 89 -76	9 56 56	9 34 38	-3* 28 28	-5* 79 65
-9 110 -100	-3* 27 11	10 87 72	-3* 30 -25	-2* 28 16	-4* 41 -15
-8 130 -122	-2* 27 -8	4, 2, L	-2 45 41	-1 32 11	-3* 41 -40
-7 24 30	1 103 -103	-1 59 56	4, 8, L	0* 36 -11	-1 84 -92
-6 245 -233	0 106 -106	-1 59 56	1 41 53	0 41 -12	
-5 181 -159	1 84 -88	-1 59 56	0 56 60	1 121 -134	
-4 55 60	2 100 -105	-7 71 -63	1 51 -48	-11* 52 36	-3 100 -93
-3 276 -282	3 100 5	-6* 23 -51	1 46 -43	-6* 124 -124	-7* 111 -111
-2 117 -130	4 71 -67	-5 65 -72	5 54 -47	-5 113 113	-6 87 76
-1 220 -221	5* 28 13	-4 344 367	4* 30 -30	-4 61 63	-5* 79 65
0 77 80	6* 26 13	-3 206 -215	5 35 25	-3* 28 28	-4* 41 -15
1 135 25	7* 24 24	-2 56 -50	-3* 30 -25	-2* 28 16	-3* 41 -40
2 338 33	8 54 31	-1 64 -61	4, 8, L	-1 32 11	-2* 41 -40
3 151 147	1 75 62	0 0 -84	1 59 56	0 36 -11	-1 84 -92
4 288 286	2 115 116	-1 85 -84	1 41 53	0 41 -12	
5 174 173	3 115 116	-1 85 -84	0 56 60	1 121 -134	
6 102 108	4 112 102	-1 66 -81	-6 58 42	-6 58 42	
7 153 160	5 102 95	-5 66 -80	-5* 24 12	-5* 22 -58	
8 32 33	6 98 95	-4 66 -80	5 40 30	-4* 22 -22	
9* 51 40	-7* 28 18	-3 61 53	-5 24 12	-3* 22 -22	
10* 24 25	-5* 69 -67	-2 306 -215	-5 24 12	-3* 22 -22	
11 32 20	-4* 28 -17	-1 206 -215	-5 24 12	-3* 22 -22	
3, 3, L	-3* 27 -8	10 24 -11	-5 24 12	-3* 22 -22	
-1* 27 26	2* 30 -14	2* 30 -14	-5 24 12	-3* 22 -22	
-2* 34 26	3* 30 -14	3* 30 -14	-5 24 12	-3* 22 -22	
-1* 28 -39	4, 3, L	4, 3, L	-5 24 12	-3* 22 -22	
0 71 76	-12 54 34	-12 54 34	-5 24 12	-3* 22 -22	
-1 43 -36	-1* 28 -10	-1 66 57	-1* 31 19	-1* 31 19	
-10 43 -40	2 39 -39	-1 66 57	0 57 52	0 57 52	
-9 48 11	3 41 29	-1 66 57	1* 32 20	1* 32 20	
-8 80 -79	4* 27 15	-1 66 57	2 69 59	2 69 59	
-7 48 37	5 64 53	-1 127 -121	2 69 59	2 69 59	
-6* 24 1	-5 26 -13	-1 127 -121	-7* 25 7	-7* 25 7	
-5 29 21	-4 359 -368	-1 127 -121	-6* 27 -7	-6* 27 -7	
-4 77 71	3, 9, L	-1 116 116	-5 64 -56	-5 64 -56	
-3 372 379	-3 116 116	-1 116 116	-4 41 -34	-4 41 -34	
-2 134 122	-2 112 -94	-1 112 -94	-3 33 -8	-3 33 -8	
0 52 55	-1 122 -94	-1 122 -94	-2 61 -56	-2 61 -56	
1 121 116	-7* 22 17	1 26 -27	-1 43 36	-1 43 36	
2 110 110	-6* 24 17	2* 26 -27	0 71 -80	0 71 -80	
3 25 4	-5 51 -44	3 47 33	1* 28 -15	1* 28 -15	
4 150 154	-4* 27 -25	4 75 74	2* 27 -8	2* 27 -8	
5* 25 25	-3* 27 -25	5 114 -111	3 42 29	3 42 29	
6* 44 44	-2* 28 -3	6 95 96	-4 24 29	-4 24 29	
-7* 45 45	-1* 28 -7	7 60 50	4, 10, L	-10 76 -60	-10 76 -60
-8* 45 45	-1* 28 -7	8 54 47	-2 35 25	-8 63 -56	-8 63 -56
-9* 45 45	-1* 27 10	9 69 56	-1* 25 -3	-7 100 -93	-7 100 -93
-8 47 50	1 34 23	10 51 45	-2 25 -3	-6 37 20	-6 37 20
-9 99 -91	2 34 -30	-	-1* 25 -3	-5 33 34	-5 33 34
-10 65 -62	3 32 -15	4, 4, L	-4 24 29	-4 24 29	
-11 69 -64	4 45 -29	-	4, 11, L	-3 41 -45	-3 41 -45
3, 4, L	5 53 40	-11 89 -82	-5 29 23	-2 60 51	
-11 43 39	3, 10, L	-10 49 -36	-5 28 -24	-1 33 -3	
-10 30 -8	-8* 30 15	-9 120 -120	-3 35 -34	0 116 121	
-9 57 -52	-7* 121 -109	-7 121 -109	-2* 33 -8	1* 34 7	
-8 60 -56	-6* 22 -21	-6 26 -27	-1 29 27	2 82 77	
-7 36 34	-5* 23 -4	-5 66 -63	-1 43 36	3 39 26	
-6 35 39	-5* 24 -21	-4 69 -60	1* 16 1	4 129 109	
-5 116 116	-3* 25 13	-3* 23 -7	-2 24 26	-5 43 -22	
-4 247 -232	-2* 25 -21	-2 146 -143	5, 1, L	-4 42 -33	
-2 161 161	-1 75 -70	-1 140 145	5, 7, L	-7 62 54	
-1 30 32	0 59 46	-1 208 203	-5 29 23	-6 63 57	
0 138 -129	1* 24 19	2 79 69	-4 28 -24	-5* 34 13	
1 62 58	2 42 43	3 266 265	-3* 29 27	-4* 34 -12	
2 42 42	3* 22 13	4 147 151	-2* 29 27	-3 61 58	
4 51 51	4 44 41	5 50 40	-1 118 113	-2* 36 38	
5 101 93	-	-	-1 148 157	-1 39 35	
6 145 -156	-2 45 -29	-	-3 36 37	-2 38 -15	
3, 5, L	5 53 40	-11 89 -82	-1 93 -92	-1 39 -15	
-1 41 16	-8 51 45	-10 93 -92	-2 38 -15	-2 38 -15	
0 80 78	4, 0, L	-7 77 -76	-3 39 -37	-3 39 -37	
-5 27 10	-4 95 82	-6 178 170	-2 93 -92	-2 38 -15	
-6 35 -36	-3 103 85	-5 39 25	-1 105 100	-1 39 35	
-5 37 33	-2 103 95	-4 95 82	-1 112 107	-1 39 35	
-4 27 33	-1* 112 98	-6* 34 31	-1 112 107	-1 39 35	
-3 28 25	0* 112 98	-5* 34 31	-1 112 107	-1 39 35	
-2 127 117	-9 76 76	2 110 95	-1 118 113	-1 39 35	
0 95 94	-8 29 29	3 114 108	-1 148 157	-1 39 35	
1 70 63	-7 73 72	4 80 -72	-1 105 100	-1 39 35	
2 41 34	-6 141 136	5* 30 -17	-1 105 100	-1 39 35	
3 38 37	-5 141 136	-4 42 -34	-1 105 100	-1 39 35	
4 44 34	-4 43 35	-3 42 -34	-1 105 100	-1 39 35	
5 77 71	0 175 -181	7 39 -34	-1 105 100	-1 39 35	
6 126 126	2 173 -295	8 86 -72	-1 105 100	-1 39 35	
7 68 70	2 174 -186	9 60 -52	-1 105 100	-1 39 35	
8 79 68	3 179 -239	-	-1 105 100	-1 39 35	
9 33 28	4 54 -52	-	-1 105 100	-1 39 35	
10 66 53	-	-	-1 105 100	-1 39 35	

as $2/m$ (C_{2h}). The systematic extinction of $h0l$ reflections when h is odd and of $0kl$ when k is odd identified the space group as $P2_1/a$ (C_{2h}^5). The unit-cell dimensions, determined from zero level Weissenberg photographs calibrated with sodium chloride, are

$$a = 7.50 \pm 0.02, b = 9.91 \pm 0.02, c = 11.677 \pm 0.005 \text{ \AA}; \\ \beta = 108.02 \pm 0.14^\circ.$$

The assumption of four molecules per unit cell leads to a calculated density of 1.365 g.cm^{-3} in good agreement with the observed value of 1.354 g.cm^{-3} measured by flotation in potassium iodide solution.

Multiple film equi-inclination Weissenberg photographs were collected, using $\text{Cu } K\alpha$ radiation, of levels Hkl for $0 \leq H \leq 5$ and hKl for $0 \leq K \leq 5$. The intensities were estimated visually with the use of standard scales. A total of 936 independent diffraction maxima were observed; within this range 197 reflections were below the minimum observed intensity M ; they were assigned the value M , and are identified with an asterisk in Table 1. The preliminary data-processing, scaling and correction for Lorentz and polarization factors were carried out with the use of programs written for the IBM 7090. Correlation of the two axes sets were made with the use of reflections common to the two sets. The weighting scheme for the least-squares refinements were chosen as $w=1/\sigma^2$, where $\sigma=0.20 I_o$ for $I_o \leq 150$ and $\sigma=0.09 I_o$ for $I_o > 150$. No corrections for absorption or extinction were made. Preliminary scale and temperature factors were estimated by Wilson's method.

Structure determination

A three-dimensional Patterson function was computed with the use of coefficients sharpened so that the average F_{hkl}^2 was independent of $\sin \theta$. The origin peak was partially removed, and the unobserved reflections were included in this calculation in order to improve the resolution of the map. Examination of the Harker plane at $y=\frac{1}{2}$ and the Harker line at $x=\frac{1}{2}, z=0$ gave three possible sets of coordinates for the Cl atom. Inspection of the area around the origin clearly indicated the positions of the six C atoms of the benzene ring, and also atom C(8), but this examination did not fix atoms 9, 10, and 11 with certainty (Fig. 1). Packing considerations and calculations of interatomic vectors for this part of the molecule then quickly eliminated the two incorrect sets of Cl coordinates. A structure factor calculation for the 480 largest reflections including contributions from Cl and 7 C atoms of the molecule gave $R=\sum|F_o|-|F_c|/\sum|F_o|=0.42$. A three-dimensional Fourier synthesis was then calculated using only those F_{hkl} 's whose signs were regarded as probably correct in the above calculation. This map showed the three additional atoms in the structure having peak heights (on a relative scale) at 49, 26 and 28, respectively, as compared with

115 for Cl and 34–42 for the C atoms. It was not possible at this stage to distinguish between the carbon atom of the CH_3 group and the oxygen atom, but packing considerations led us to assign atom 10 as O and atom 11 as C. This choice was independently tested later, as described below.

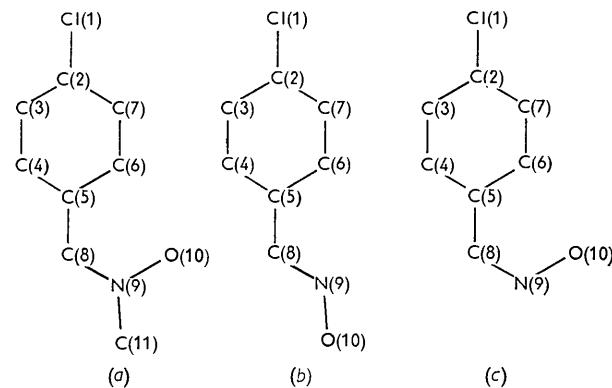


Fig. 1. Numbering diagram for the three oximes discussed in this paper. (a) *N*-Methyl-*p*-chlorobenzaldoxime, (b) 'sym'-*p*-chlorobenzaldoxime and (c) 'anti'-*p*-chlorobenzaldoxime

All observed reflections were then introduced. Three cycles of least-squares refinement with individual isotropic temperature factors reduced the disagreement index to $R=0.184$. Four H atoms on the benzene ring were introduced at expected positions, and were included in the refinements as fixed atoms. Two cycles of anisotropic refinement then yielded $R=0.097$, $R_{F2}=\sum|F_o^2-F_c^2|/\sum F_o^2=0.159$, and $R_{wF2}=\sqrt{(\sum w|F_o^2-F_c^2|^2)/\sum w F_o^4}=0.25$. All distance parameter shifts were less than their respective standard deviations in the final cycle of this full matrix refinement. Only one coordinate shifted by as much as 0.007 \AA , and the average absolute shift of coordinates was less than 0.002 \AA in this final cycle of refinement. The final shifts of diagonal thermal parameters (β_{ii}) were less than the standard deviations of the β_{ii} by a factor of four.

This low disagreement factor suggested that our choice of the identities of atoms 10 and 11 had been correct. Furthermore, the individual atomic thermal parameters had behaved normally throughout the refinements. Nevertheless, we made a further impartial test of the assignment of atom 10 as O and atom 11 as C (Table 2). Fixed atom contributions were calculated for atoms 1 to 9, and atoms 10 and 11 were each given the form factor of oxygen. The individual isotropic temperature factors were then refined. It would thus be expected that a correct choice of the identities of atoms 10 and 11 would lower the disagreement factor and improve the temperature factor, while the wrong choice would have just the opposite effect. Indeed this was the case: choice II increased the refined R and gave an ab-

Table 2. Test of the assignment of atoms 10 and 11

Condition	Isotropic temperature factor		$R = \sum F_o - F_c / \sum F_o $	
	Initial	Final	Initial	Final
I Atom 10=oxygen Atom 11=oxygen	4.0	5.06	0.178	0.111
	4.0	8.93		
II Atom 10=carbon Atom 11=oxygen	5.06	1.75	0.148	0.134
	8.93	10.42		
III Atom 10=oxygen Atom 11=carbon	5.06	5.24	0.118	0.103
	8.93	5.10		

normally large thermal parameter for atom 11, while choice III resulted in reasonable temperature factors and a decrease in the refined value of R (Table 2).

We then felt confident that the structure was determined correctly, and that the configuration was that of the 'anti' oxime. Confirmatory evidence of an

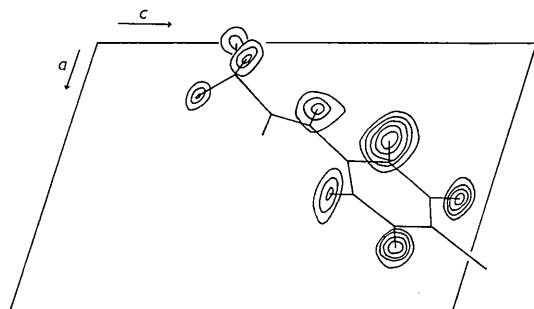


Fig. 2. Composite map on (010) of the electron density through the centers of the hydrogen atoms from the final difference map. A skeleton of the molecule is superposed. The contours are at intervals of $0.1 \text{ e.} \text{\AA}^{-3}$, starting at $0.3 \text{ e.} \text{\AA}^{-3}$.

independent kind was obtained from a final three-dimensional difference electron density map that showed all the H atoms of the structure. A composite projection of this map is shown in Fig. 2. No false peaks higher than $0.2 \text{ e.} \text{\AA}^{-3}$ occurred in this map. Introduction of these eight H atoms, followed by one cycle of isotropic least-squares refinement with the use of observed reflections only lowered R to 0.085, R_{F^2} to 0.143, and R_{wF^2} to 0.224. In Table I we list the observed and calculated structure factors including unobserved reflections which were not used in the least-squares refinement.

Discussion of the structure

The final structure parameters are given in Tables 3 and 4, and a summary of the disagreement factor R is given in Table 5. The molecular packing as viewed along the b axis is shown in Fig. 3 and the numbering diagram in Fig. 1(a). Bond distances and angles with their standard deviations were computed using the full variance-covariance matrix from the

Table 3. Final atomic parameters and their standard deviations for *N*-methyl-p-chlorobenzaldoxime

Atom	x	y	z	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$
Cl(1)	0.8464	0.1341	1.0410	0.0004	0.0002	0.0002
C(2)	0.6914	0.1122	0.8936	0.0010	0.0006	0.0005
C(3)	0.5834	-0.0011	0.8675	0.0011	0.0006	0.0006
C(4)	0.4617	-0.0198	0.7521	0.0010	0.0006	0.0005
C(5)	0.4499	0.0780	0.6621	0.0008	0.0005	0.0005
C(6)	0.5675	0.1907	0.6930	0.0009	0.0006	0.0005
C(7)	0.6883	0.2081	0.8093	0.0010	0.0006	0.0006
C(8)	0.3133	0.0468	0.5452	0.0009	0.0006	0.0005
N(9)	0.2685	0.1280	0.4522	0.0008	0.0004	0.0004
O(10)	0.3439	0.2434	0.4480	0.0007	0.0004	0.0004
C(11)	0.1196	0.0849	0.3397	0.0011	0.0007	0.0006
B^*						
H(3)	0.604	-0.065	0.934	2.2		
H(4)	0.361	-0.097	0.724	0.6		
H(6)	0.553	0.257	0.629	3.1		
H(7)	0.777	0.285	0.827	1.3		
H(8)	0.244	-0.053	0.536	2.8		
H(11')	0.075	-0.013	0.337	6.5		
H(11'')	0.175	0.100	0.268	5.6		
H(11''')	0.016	0.155	0.327	5.4		

Scale factor: 2.208 ± 0.009 †

* Little significance can be attached to the isotropic temperature factors B given for the H atoms. In general the standard errors were comparable to the B 's themselves.

† The scale factor relates the absolute scale of the calculated F 's and the relative scale of the observed F 's, $|F_{\text{rel}}| = |F_{\text{abs}}| \cdot s$.

Table 4. Final anisotropic thermal parameters for *N*-methyl-*p*-chlorobenzaldoxime

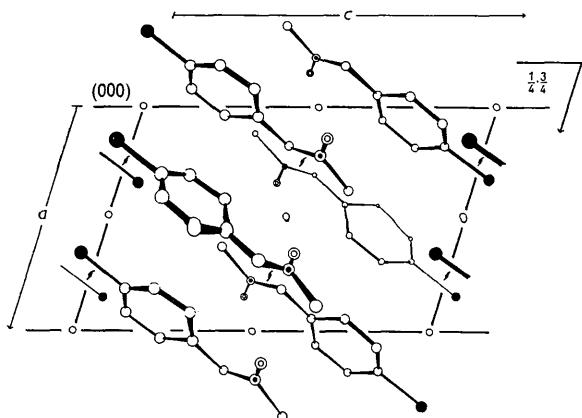
Atom	β_{11}	β_{22}	β_{33}	β_{12}	β_{13}	β_{23}
Cl(1)	0.0395	0.0176	0.0085	-0.0017	-0.0017	-0.0010
C(2)	0.0269	0.0122	0.0077	0.0009	0.0028	-0.0006
C(3)	0.0337	0.0104	0.0096	0.0039	0.0063	0.0006
C(4)	0.0287	0.0096	0.0107	-0.0016	0.0065	-0.0003
C(5)	0.0195	0.0094	0.0093	0.0020	0.0059	0.0003
C(6)	0.0207	0.0093	0.0096	-0.0010	0.0050	0.0012
C(7)	0.0241	0.0100	0.0114	-0.0009	0.0043	-0.0007
C(8)	0.0253	0.0097	0.0095	0.0020	0.0062	-0.0002
N(9)	0.0233	0.0100	0.0095	0.0011	0.0062	-0.0002
O(10)	0.0258	0.0122	0.0113	-0.0015	0.0045	0.0027
C(11)	0.0303	0.0159	0.0088	0.0030	0.0025	0.0015

least-squares refinement and are given in Table 6, and shown in Figs. 4 and 5. We feel that the reported standard deviations should be multiplied by a factor of 2 or 3 to obtain a reasonable level of significance.

Table 5. Final values of $R = \Sigma |F_o| - |F_c| / \Sigma |F_o|$ for *N*-methyl-*p*-chlorobenzaldoxime

hkl	R	Range of $\sin \theta$	R
All orders	0.089	0.00 to 0.40	0.074
h even	0.088	0.40 to 0.50	0.065
h odd	0.090	0.50 to 0.60	0.060
k even	0.088	0.60 to 0.65	0.077
k odd	0.090	0.65 to 0.70	0.096
l even	0.086	0.70 to 0.75	0.105
l odd	0.092	0.75 to 0.80	0.121
$k+l$ even	0.087	0.80 to 0.85	0.136
$k+l$ odd	0.091	0.85 to 0.90	0.149
$h+l$ even	0.093	0.90 to 1.00	0.177
$h+l$ odd	0.086		
$h+k$ even	0.083		
$h+k$ odd	0.096		
$h+k+l$ even	0.094		
$h+k+l$ odd	0.085		

Overall disagreement $R = 0.085$ for F_o only, and $R = 0.108$ if 197 unobserved F 's are included.

Fig. 3. The structure of *N*-methyl-*p*-chlorobenzaldoxime viewed along the b axis. Black circles represent Cl atoms, open circles C atoms, circles with a dot N atoms and double circles O atoms. The origin is at (000).Table 6. Intramolecular bonds and angles in *N*-methyl-*p*-chlorobenzaldoxime

Atoms	Distance Å	Standard deviation*
Cl(1)-C(2)	1.768	± 0.007
C(2)-C(3)	1.363	0.009
C(3)-C(4)	1.388	0.009
C(4)-C(5)	1.411	0.008
C(5)-C(6)	1.401	0.008
C(6)-C(7)	1.391	0.008
C(7)-C(2)	1.363	0.008
C(5)-C(8)	1.464	0.008
C(8)-N(9)	1.309	0.007
N(9)-O(10)	1.284	0.006
N(9)-C(11)	1.501	0.008
C(3)-H(3)	0.98	
C(4)-H(4)	1.06	
C(6)-H(6)	0.98	
C(7)-H(7)	0.99	
C(8)-H(8)	1.11	
C(11)-H(11')	1.02	
C(11)-H(11'')	1.05	
C(11)-H(11''')	1.01	
Atoms	Angle	Standard deviation*
Cl(1)-C(2)-C(3)	119.1°	$\pm 0.5^\circ$
C(2)-C(3)-C(4)	119.6	0.5
C(3)-C(4)-C(5)	120.2	0.5
C(4)-C(5)-C(6)	117.8	0.5
C(5)-C(6)-C(7)	121.3	0.5
C(6)-C(7)-C(2)	118.6	0.6
C(7)-C(2)-C(3)	122.5	0.6
Cl(1)-C(2)-C(7)	118.5	0.5
C(6)-C(5)-C(8)	127.8	0.5
C(4)-C(5)-C(8)	114.5	0.5
C(5)-C(8)-N(9)	124.7	0.5
C(8)-N(9)-O(10)	125.2	0.5
C(8)-N(9)-C(11)	118.8	0.5
O(10)-N(9)-C(11)	115.9	0.5

* The standard deviations are calculated using the full variance-covariance matrix from the least-squares refinement. The values should be multiplied by 2 or 3 for significant uncertainty.

The Cl atom and the benzene ring are coplanar to within 0.01 Å. C(8) is 0.04 Å from the approximate plane through Cl and the ring ($X - 0.58Y - 0.30Z + 1.67 = 0$), while N(9), O(10), and C(11) are removed 0.17, 0.22, and 0.24 Å, respectively, from this plane.

The oxime group has caused a slight displacement of the C(5)-C(8) bond that is significantly out of

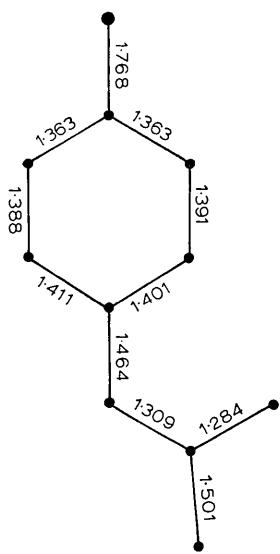


Fig. 4. Bond lengths in *N*-methyl-*p*-chlorobenzaldoxime.

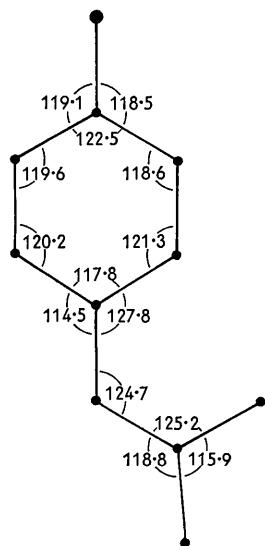


Fig. 5. Bond angles in *N*-methyl-*p*-chlorobenzaldoxime.

the line bisecting the benzene ring: the angle C(6)-C(5)-C(8) is 127.8°. The C-C bonds of the ring are essentially normal; the average value for the C-C distances of this ring is 1.386 Å and the average of those angles not involving H atoms of the ring is 120.0°. The C(5)-C(8) distance is slightly larger in the *N*-methyl compound than in either '*syn*' or '*anti*'-*p*-chlorobenzaldoxime. The N(9)-O(10) distance of 1.284 Å, remarkably shorter than that reported for non-alkylated oximes, indicates multiple bonding

of a larger degree than that in the oximes themselves. The distance N(9)-C(11)=1.501 Å is only slightly longer than the value reported for 3-covalent nitrogen (1.472 Å). Finally, the bond distances N(9)-O(10)=1.284 Å and N(9)-C(11)=1.501 Å provide additional strong support for the assignment of the '*anti*' configuration of *N*-methyl-*p*-chlorobenzaldoxime.

Short intramolecular distances are C(5) ··· O(10)=2.889, C(6) ··· O(10)=2.883, C(8) ··· O(10)=2.302, C(11) ··· O(10)=2.364, C(8) ··· C(11)=2.421 and C(6) ··· C(8)=2.417 Å, all ± 0.008 Å. The first two distances may be compared to the non-bonded distance of $1.40 + 1.55 = 2.95$ Å which might be expected from van der Waals radii of O and C. No abnormally short intermolecular contacts are present. The closest Cl ··· Cl distance is 3.66 Å, and the closest intermolecular C ··· O distance is 3.24 Å. The angle between normals to the benzene ring and the C(8)N(9)O(10) plane is 5° 37'. This small twist is distributed between the C(5)-C(8) and C(8)-N(9) bonds. The angle between the normals to the C(5)C(8)N(9) plane and the benzene plane indicates that the C(8)N(9) bond is twisted around C(5)C(8) by 6° 4'. Also, the angle between the normals to the benzene ring and the average plane defined by C(5)C(8)N(9)O(10) is 4° 27'. These twists place atoms C(8), N(9), O(10) and C(11) slightly above the plane of the ring, provided that we define the positive direction by a vector from the origin of the unit cell toward the plane. If a coordinate Z is defined perpendicular to the crystallographic $x=X$ and $y=Y$ axes, the equation of the plane of the ring is

$$X - 0.5749 Y - 0.3026 Z + 1.6709 = 0$$

where the units are in Å. The distances of atoms from this plane are C(8), 0.04 Å; N(9), 0.17 Å; O(10), 0.22 Å; and C(11), 0.24 Å.

Refinement of '*syn*'- and '*anti*'-*p*-chlorobenzaldoximes

The three-dimensional structure determination of '*syn*'-*p*-chlorobenzaldoxime, reported by Jerslev (1957), was carried out by means of three-dimensional Fourier syntheses. It was felt that these original data merited further refinement so that comparison could be made between the structures of '*syn*'-*p*-chlorobenzaldoxime, *N*-methyl-*p*-chlorobenzaldoxime, and an *O*-methyl derivative of a closely related oxime to be studied in the near future. The present refinement of '*syn*'-*p*-chlorobenzaldoxime was done by the least-squares method starting from the values reported by Jerslev (1958). The weighting scheme is as follows: $w=1/\sigma^2$, where $\sigma=F_o^2$ for $F_o^2 \leq 4.F_{\min}^2$ and $\sigma=0.10.F_o^2$ for $F_o^2 > 4.F_{\min}^2$. Two cycles of three-dimensional least-squares refinement in which individual isotropic temperature factors were assumed reduced $R=\sum||F_o|-|F_c||/\sum|F_o|$ to 0.151. Introduction of the four

THE STRUCTURE OF *N*-METHYL-*p*-CHLOROBENZALDOXIMETable 7. List of observed and calculated structure factors for 'syn'-*p*-chlorobenzaldoximeThe three columns of each group contain from left to right the values of *l*, 10 F_o and 10 F_c .The values of *l* followed by an asterisk indicate reflections below the observable limit

		0, 0, L	-7, 1, L	-2, 1, L	1, 1, L	5, 1, L
-7, 0, L						
2	33	27	24	123	123	1
4	49	51	26	55	54	2
6	17	-21	23	106	-110	3
8	44	45	30	24	20	4
10	42	-13	32	34	-35	5
12	27	30	12	75	69	6
14*	27	0	12	-50	-13	7
16	23	-41	0	1, 0, L		8
-6, 0, L				-6, 1, L		
2	33	27	0	197	-398	9
4	49	51	2	505	-122	10
6	17	-21	4	744	885	10
8	43	51	8	327	412	11
10	43	157	12	290	-312	12
12	11	20	14	324	41	13
14	19	-16	16	35	30	14
16	0	-17	16	112	-106	15
18	43	51	20*	4	10	16*
20	29	-26	22	177	-171	17
22	29	30	24	51	47	18
-5, 0, L				26	13	19
2	133	157	30	70	71	20
4	119	-14	18	11	17	21
6	53	63	20	25	-13	22
8	33	17	20	11	3	23
10	262	-332	20	11	-5, 1, L	24
12	78	81	7	267	-291	25*
14	30	-41	8	62	-50	27
16	20	24	10	59	61	27
18	69	74	12	243	-254	28
20	75	-65	14	225	227	29
22	72	57	16	113	96	30
-4, 0, L				30	181	31
2	201	229	20	60	67	32
4	101	-111	22	25	-30	33
6	198	234	24	9	41	34
8	95	-81	26	15	20	35
10	96	87	30	34	-29	36
12	147	-171	3, 0, L	14	40	37
14	21	38	0	114	-122	38
16	62	-66	2	163	191	39
18	82	-85	4	286	301	40
20	105	-77	6	420	-454	41
22	59	60	10	209	-219	42
-3, 0, L				22	22*	43
2	91	99	14	277	279	44
4	191	-255	16	392	-413	45
6	211	234	20	46	48	46
8	281	-237	22	31	30	47
10	102	105	24	123	123	48
12	448	457	26	25	-30	49
14	145	-143	28	17	18	50
16	140	147	4, 0, L	5	103	51
18	80	-92	6	190	-207	52
20	37	-27	0*	0	-6	53
22	175	161	2*	0	12	54
24	72	-72	2	39	44	55
26	59	-63	6	103	-145	56
28	75	-75	8	122	126	57
-2, 0, L				4-	1, L	
2	113	166	12	51	-52	
4	177	-165	18	174	179	
6	305	300	20	115	-107	
8	247	-219	22	118	115	
10	202	220	24	61	-49	
12	167	-167	26	80	-66	
14	157	-146	14	103	106	
16	292	255	16	100	-108	
18	243	-218	18	17	167	
20	0	127	22	7	171	
22	25	-35	24*	0	152	
24	125	-137	26	9	219	
26	202	193	28	1	-171	
28*	-0	-6	8	82	-70	
30	23	20	10	115	-79	
-1, 0, L				-3, 1, L		
2	92	-93	14	46	50	
4	239	-276	16	58	-52	
6	56	-66	18	53	77	
8	334	-462	20	150	-22	
10	152	159	22	125	126	
12	163	-161	24	61	-49	
14*	112	125	26	80	-66	
16	12	32	0	110	-111	
18	152	-165	2	113	119	
20	145	150	4	119	-134	
22	97	-94	6	65	-74	
24*	0	7	8	73	92	
26	41	41	10*	59	69	
28	50	-50	12	14	116	
30	83	76	15	16	63	
32	39	-23	18	14	3	
0, 0, L				7, 0, L		
2	119	-150	0	17	-25	
4	334	-360	2	20	20	
6	210	198	4	55	-51	
8	508	-534	6	42	38	
10	501	465	8	34	-34	
12	247	-255	10	67	-63	
14	242	-211	12	20	17	
16	118	-127	24	27	33	
18	106	-105	26	28	85	
20	138	129	28	29	25	
22	128	-133	30	33	-54	
0, 0, L				1, 1, L		
2	33	27	0	197	-398	
4	49	51	2	505	-122	
6	17	-21	3	53	50	
8	43	51	5	14	13	
10	42	-13	7	10	75	
12	27	30	9	12	75	
14	27	0	10	12	75	
16	23	-41	12	12	75	
18	119	-14	14	12	75	
20	18	20	15	12	75	
22	175	161	17	12	75	
24	72	-72	18	12	75	
26	59	-63	20	12	75	
28	75	-75	22	12	75	
-1, 0, L				2, 1, L		
2	33	27	1	102	-90	
4	49	51	2	34	-38	
6	17	-21	4	20	26	
8	43	51	5	14	13	
10	42	-13	7	12	12	
12	27	30	9	12	12	
14	27	0	10	12	12	
16	23	-41	12	12	12	
18	119	-14	14	12	12	
20	18	20	15	12	12	
22	175	161	17	12	12	
24	72	-72	18	12	12	
26	59	-63	20	12	12	
28	75	-75	22	12	12	
-5, 0, L				6, 1, L		
2	33	27	1	197	-398	
4	49	51	2	505	-122	
6	17	-21	3	53	50	
8	43	51	5	14	13	
10	42	-13	7	12	12	
12	27	30	9	12	12	
14	27	0	10	12	12	
16	23	-41	12	12	12	
18	119	-14	14	12	12	
20	18	20	15	12	12	
22	175	161	17	12	12	
24	72	-72	18	12	12	
26	59	-63	20	12	12	
28	75	-75	22	12	12	
-4, 0, L				7, 1, L		
2	201	229	1	84	92	
4	101	-111	2	65	-59	
6	198	234	4	243	-254	
8	95	-81	6	15	-29	
10	96	87	8	14	-34	
12	147	-171	10	20	-35	
14	21	38	12	20	-41	
16	21	38	14	20	-46	
18	18	21	16	20	-51	
20	37	-27	18	20	-56	
22	22	22	20	20	-62	
24	22	22	22	20	-67	
26	22	22	22	20	-72	
28	22	22	22	20	-77	
-3, 0, L				8, 1, L		
2	91	99	1	84	92	
4	191	-255	2	65	-85	
6	211	234	4	243	-254	
8	281	-237	6	15	-29	
10	102	105	8	14	-34	
12	448	457	10	20	-35	
14	145	-143	12	20	-41	
16	140	147	14	20	-46	
18	80	-92	16	20	-51	
20	37	-27	18	20	-56	
22	22	22	20	20	-62	
24	22	22	22	20	-67	
26	22	22	22	20	-72	
28	22	22	22	20	-77	
-2, 0, L				9, 1, L		
2	113	166	1	14	96	
4	177	-165	2	98	-96	
6	305	300	4	126	-133	
8	247	-219	6	24	25	
10	202	220	8	19	20	
12	167	-167	10	106	-106	
14	157	-146	12	14	-25	
16	292	255	14	21	*0	
18	243	-218	16	17	24	
20	0	127	18	18	24	
22	25	-35	24*	0	14	
24	125	-137	26	11	-12	
26	202	193	28	81	-68	
28*	-0	-6	8	0	3	
30	23	20	10	115	-79	
-1, 0, L				-3, 1, L		
2	14	43	1	211	171	
4	16	58	2	317	347	
6	56	-66	3	47	48	
8	334	-462	4	72	68	
10	152	159	5	52	53	
12	163	-161	6	239	-208	
14*	112	125	7	243	213	
16	152	-165	8	10	145	
18	152	-165	10	116	-143	
20	145	150	12	34	32	
22	97	-94	13	32	-26	
24*	0	7	14	32	-26	
26	41	41	15	66	-111	
28	50	-50	16	59	69	
30	83	76	18	14	3	
32	39	-23	0	110	-111	
0, 0, L				7, 0, L		
2	119	-150	0	17	-25	
4	334	-360	2	20	20	
6	210	198	4	55	-51	
8	508	-534	6	42	38	
10	501	465	8	34	-34</td	

Table 7 (cont.)

		0, 2, L	4, 2, L	-4, 3, L	0, 3, L	4, 2, L						
-4, 2, L												
9	63	55	6	50	49	10	113	115	11*	0	-11	
10	37	28	8	207	167	1	23	-5	12	62	-76	
11	96	-90	9	117	129	2*	0	-13	13	35	-95	
12*	0	-0	10*	0	-131	-121	3*	39	37	14	45	-44
13	134	11	240	219	4*	3	19	-22	13	299	259	
14	66	-70	12	29	16	5	135	135	14	158	-148	
15	44	-39	13*	174	-160	6	54	47	15	188	-159	
16	20	-14	14	37	-41	7	241	-263	16	145	-144	
17	92	91	15	59	66	8	83	75	17	76	-63	
18	23	20	16	59	66	9	81	83	18*	1	-19	
19*	0	13	17*	53	58	10*	0	11	19	75	-72	
20*	0	-2	18*	0	-14	11	52	41	20*	0	21	
21	88	-81	19	173	-158	12	35	30	21	48	-44	
22	43	-31	20	64	-57	13	28	-41	22	32	29	
23	67	57	21	166	149	14	40	-46	23	71	62	
24*	0	-1	22	64	-55	15	43	36	24	56	-55	
25	86	-71	23	59	-55	16	43	-38	25	52	-50	
26	21	11	24	28	-34	17	64	-65	26*	5	5	
			25	41	43	18	11	-15	18*	0	-16	
-3, 2, L			26	34	-32	19	11	-15	20	63	-52	
1*	0	8	27	23	29	20*	0	-11	21*	0	-13	
2	32	-24	28	35	31	22	34	-27	22	48	44	
3	201	149	29	56	-49	23	37	23	-3, 3, L	0*	0	
4	205	149	30	24	18	24	38	26	1	122	12	
5	238	-261		1, 2, L		5, 2, L	1*	0	2	110	110	
6	171	131					2	-18	3	140	122	
7	119	109	0*	0	20	3	45	38	4	116	-89	
8	97	80	1	236	223	4	64	-55	5	85	70	
9	123	-107	2	41	55	5	71	67	7	165	149	
10	111	105	3	323	-307	6	72	-54	8	67	-73	
11	79	-74	4	131	-121	7	77	-75	9	95	-88	
12	28	-35	5*	0	-21	8	121	96	10	73	69	
13	113	95	6	100	91	9	275	235	11	81	-80	
14	30	35	7	189	-159	10	84	-67	12	150	139	
15	181	-160	8	121	-122	11*	0	11	13	29	-35	
16	28	23	9	123	-113	12*	0	-11	14	29	-38	
17	110	94	10	74	79	13*	0	-20	15*	0	-26	
18	62	-50	11	347	309	14*	0	-27	16	88	-26	
19*	0	-24	12	157	156	15*	0	-9	17	5	4	
20*	0	-2	13	129	-122	16	28	29	18	59	-59	
21	30	24	14*	0	-14	17	73	-74	19	53	-51	
22*	0	-33	15	192	167	18*	0	-17	20	66	60	
23	83	80	16*	0	-8	19	53	50	21	58	57	
24*	0	-3	17*	0	-8	20*	0	-30	22	37	-28	
25	54	-53	18*	0	-8	21	60	-57	23	49	-46	
26*	0	-3	19	132	-129	22	51	53	24*	0	-32	
27	62	64	20	30	39	23*	0	7	25	64	69	
			21	142	132	24	41	-51	26	84	71	
-2, 2, L			22	65	54	25	23	29	27	88	71	
1	150	104	23*	0	-21	6, 2, L	-2, 3, L		1*	0	-13	
2	307	-247	24	46	-39				2	79	-76	
3	329	249	25*	0	-9	0*	0	-12	3*	0	-9	
4	347	-364	26	49	-50	1	96	-94	4	44	43	
5	553	-486	27	31	-32	2	37	-33	5	52	-23	
6	173	-145	28*	0	-15	3	25	25	6	48	-49	
7	301	216	29	25	-30	4	18	15	7	115	-115	
8	152	136		2, 2, L		5	35	-34	8	53	-54	
9	73	71				6	34	-32	9	0	-50	
10	88	72	0	121	-111	7	0*	0	10	119	89	
11	39	-37	1	243	197	8*	0	-2	12	58	-55	
12	37	-33	2	48	50	9*	0	0	13	97	-93	
13	112	89	3	314	-251	10	0	-1	14	120	117	
14	64	64	11	104	-85	11	54	-46	15	102	-82	
15	189	-169	12	126	127	12	29	-22	16	37	39	
16	86	86	13	122	-104	13	30	-25	17	123	115	
17*	0	6	14	44	-37	14	20	20	18*	0	-15	
18	118	102	15*	0	-17	15*	20	-3	19	54	-53	
19	100	-85	16*	0	-9	16*	0	3	20	66	-61	
20*	0	-9	17	102	102	17*	0	-13	21*	0	-16	
21	52	-50	18*	0	-17	18	86	88	22*	0	-18	
22*	0	-17	19	33	-13	19*	0	-26	23*	0	-38	
23*	0	-17	20	22	-20	20	96	-94	24*	0	-31	
24*	0	14	21*	0	-17	21	30	21	25*	0	-35	
25	45	-49	22*	0	-12	22	66	53	26*	0	-18	
26*	0	-12	23	112	106	23*	0	-10	27*	0	-25	
27	67	63	24*	0	-12	24*	37	-30	28*	0	-28	
28*	0	-2	25	17	113	25	31	-24	29	25	31	
29	39	-40	26*	0	-18	26*	46	43	30*	0	-36	
			27	124	122	27*	21	17	31*	0	-38	
-1, 2, L			28*	0	4+	-6, 3, L	-1, 3, L		3, 3, L	0*	0	
1	265	229	29	54	-49	1	28	32	2, 3, L	16	37	
2	410	-361	30*	167	-157	2*	0	-3	3*	0	5	
3	201	139	25	90	88	3	55	-55	4	44	-46	
4	87	87	26	63	61	4	28	33	5	52	-51	
5*	0	36	27*	0	-10	5	37	34	6	59	-56	
7	522	467	28	31	-27	6	27	-24	7	61	-59	
8	316	-241		3, 2, L		7	32	-25	8	67	-67	
9	555	-512				8	58	61	9*	0	-114	
10	336	-225	9	25	9	9*	0	-11	10	46	49	
11	85	85	10	20	11	10	27	-26	11	103	94	
12	79	-83	11	73	-73	11	50	42	12	39	-41	
13	73	61	12	167	-172	12	46	-34	13	85	-87	
14	105	-96	13	111	-104	13	28	-21	14	104	-104	
15	64	-60	14	107	157	14	35	28	15	126	-123	
16	347	354	15	49	-31	15	27	-24	16	61	-56	
17	105	-98	16	139	-121	16	53	-50	17	87	-86	
18	126	-110	17	127	-124	17*	0	-8	18*	0	0	
19	126	-110	18*	9	97	18*	0	-8	19*	56	-52	
20*	0	-28	19	60	-64	20*	0	-7	20	49	54	
21*	0	-28	20	40	32	21	60	-62	21	50	51	
22	75	63	21	12	25	22	34	35	22	27	-22	
23	30	-23	22	13	25	23	55	52	23	31	-25	
24	76	79	23	15	24	24	56	-53	24	105	100	
25	30	-33	24	22	-20	25	88	-93	25	31	-25	
26	25	-20	25	104	92	26*	0	-33	26	105	100	
27	33	42	16	46	43	27	20	16	27	48	46	
28*	0	-10	17	75	-69	28*	29	-22	0, 3, L	0*	0	
29	38	-41	18*	0	4	11	28	23	1	62	-69	
			19	22	24	12	49	48	2	110	-121	
0, 2, L			20*	0	-5	13	21	17	3*	0	-27	
1*	0	-27	21	23	26	14	44	43	4	48	45	
2	268	253	22*	0	-17	15	45	-41	5	117	140	
3	106	105	24*	0	-5	16	48	-39	6	142	166	
4	86	91	25	37	34	17	59	56	7	33	-37	
5	139	136	26*	0	-3	18	31	27	8	94	77	
			19	17	-38	20	38	-36	9	42	-52	
0, 3, L			21	17	-38	21	38	-36	9	42	-52	
1*	0	-10	18*	0	4	10	29	-22	10*	0	-14	
2	206	205	20*	0	-5	11	28	23	1	52	59	
3	94	-71	21*	0	5	12	45	-41	2	30	-25	
4	86	91	22*	0	-5	13	48	-39	3	52	-53	
5	139	136	23*	0	-5	14	53	56	4	48	-45	

Table 7 (cont.)

-2, 4, L		3, 4, L		0, 5, L	
11*	0	2	0	34	-30
12	59	64	23	27	27
13	41	-31	22	-27	-22
14	22	35	23	-31	57
15*	0	-23	27	-13	53
16	84	-87	77	32	33
17*	0	4	12	125	-46
18*	95	96	7*	14	-27
19	23	-13	8	146	-69
20	38	-43	9	102	9
21	31	42	10*	0	24
22**	0	5	11*	56	-31
23**	0	-16	11*	0	23
-1, 4, L		13*	0	13	-13
1	82	-77	14*	0	21
2*	119	119	15	97	-27
3*	0	13	15	17	17
4*	49	49	16	33	17
5*	0	9	17*	77	31
6*	0	10	13	30	11
7*	82	-77	19	23	25
8	7*	0	20*	0	30
9	76	65	4, 4, L		1, 5, L
10	54	-92	0	54	34
11*	0	3	1*	0	34
12	61	51	2*	43	-32
13*	0	5	2*	45	-12
14*	0	-24	3*	45	124
15*	0	12	5*	29	125
16	50	-56	6	37	43
17	73	-73	7	29	37
18	71	67	8	32	14
19*	0	2	9*	0	16
20	40	-41	10	43	0
21*	0	-15	11	37	19
22	52	53	12	24	23
23*	0	-13	13*	0	27
0, 4, L		14*	0	16	-30
1	201	-191	15	27	61
2	0	37	16	-24	-77
3	186	180	5, 4, L	0	14
4*	71	62	0	51	24
5*	0	5	1*	0	-5
6	0	-20	2*	-1	2
7	76	-70	3*	33	-12
8	135	111	4*	-30	23
9	210	172	5*	0	28
10	41	-52	6*	0	23
11	116	-104	7	44	45
12	91	82	8	50	53
13	35	-32	9*	0	95
14	34	39	10	-41	-125
15*	63	-56	11	54	56
16	0	3	12	51	56
17	22	21	13	30	27
18	45	48	14*	0	4
19*	50	49	15	27	15*
20	79	-69	16	-41	0
21*	0	18	17	28	-13
22*	0	-8	18	55	35
24	60	-44	19	-46	44
0, 4, L		20	16	2, 5, L	0
1	81	-64	21	5, 4, L	64
2*	0	-23	22	54	-77
3*	139	130	23	57	14
4	0	26	24	58	24
5*	102	-100	25	58	-36
6	-0	-5	26*	0	34
7	96	103	27	59	34
8	30	22	28	60	34
9	91	95	29*	0	23
10	113	-97	30	22	59
11	79	-72	10*	21	50
12	179	157	11*	0	15
13	64	-61	12*	0	16
14	41	-27	13	23	56
15*	0	8	14	14	32
16*	0	18	15	5	32
17*	0	-26	16	-8	-8
18*	0	-13	17*	0	-13
19	31	30	1, 5, L	0*	0
20	56	-55	2	32	10
21*	0	10	21	-33	29
22	56	-44	22	15	29
23*	0	-10	23	58	10
-3, 5, L		24	19*	0	10
1	81	-64	25	54	-13
2	0	23	26*	0	29
3*	139	130	27	57	29
4	0	26	28	58	29
5*	102	-100	29*	0	10
6	-0	-5	30	59	10
7	96	103	31*	0	23
8	30	22	32	60	23
9	91	95	33*	0	39
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 5, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0
-2, 5, L		25	19*	0	10
1	81	-64	26*	0	-12
2	0	23	27	59	12
3*	139	130	28*	0	-12
4	0	26	29	60	12
5*	102	-100	30*	0	-35
6	-0	-5	31	59	6
7	96	103	32*	0	14
8	30	22	33*	0	14
9	91	95	34*	0	14
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 5, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0
-3, 5, L		25	19*	0	10
1	81	-64	26*	0	-12
2	0	23	27	59	12
3*	139	130	28*	0	-12
4	0	26	29	60	12
5*	102	-100	30*	0	-35
6	-0	-5	31	59	6
7	96	103	32*	0	14
8	30	22	33*	0	14
9	91	95	34*	0	14
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 5, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0
-1, 6, L		25	19*	0	10
1	81	-64	26*	0	-12
2	0	23	27	59	12
3*	139	130	28*	0	-12
4	0	26	29	60	12
5*	102	-100	30*	0	-35
6	-0	-5	31	59	6
7	96	103	32*	0	14
8	30	22	33*	0	14
9	91	95	34*	0	14
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 6, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0
-1, 6, L		25	19*	0	10
1	81	-64	26*	0	-12
2	0	23	27	59	12
3*	139	130	28*	0	-12
4	0	26	29	60	12
5*	102	-100	30*	0	-35
6	-0	-5	31	59	6
7	96	103	32*	0	14
8	30	22	33*	0	14
9	91	95	34*	0	14
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 6, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0
-1, 6, L		25	19*	0	10
1	81	-64	26*	0	-12
2	0	23	27	59	12
3*	139	130	28*	0	-12
4	0	26	29	60	12
5*	102	-100	30*	0	-35
6	-0	-5	31	59	6
7	96	103	32*	0	14
8	30	22	33*	0	14
9	91	95	34*	0	14
10	113	-97	10*	21	39
11	79	-72	11*	0	39
12	179	157	12*	0	10
13	64	-61	13	23	56
14	41	-27	14	14	32
15*	0	8	15	5	32
16*	0	18	16	-8	-8
17*	0	-26	17*	0	-13
18*	0	-13	1, 6, L	0*	0
19	31	30	2	32	10
20	56	-55	21	-33	29
21*	0	10	22	15	29
22	56	-44	23	58	10
23*	0	-10	24	19*	0

H atoms on the benzene ring as fixed atoms, followed by anisotropic variation of thermal parameters for the heavier atoms gave, after one cycle, the final $R = 0.116$, $R_{F^2} = \sum |F_o^2 - F_c^2| / \sum F_o^2 = 0.20$ and $R_{wF^2} = \sqrt{\sum w|F_o^2 - F_c^2|^2 / \sum w|F_o|^4} = 0.30$ for the 1238 observed diffraction maxima. When the 338 unobserved reflections were included the value of R became 0.151. The full matrix was used in all least-squares refinements, and all positional shifts were less than their standard deviations in the final refinement. For example, in the final refinement of '*syn*'-*p*-chlorobenzaldoxime the largest parameter shift was 0.0010 Å, and the average absolute shift was 0.0004 Å.

The observed and calculated structure factors are given in Table 7, where unobserved values are indicated by an asterisk. A summary of the disagreement index R (observed reflections only) is given in Table 8, and the final atomic parameters and their standard deviations are given in Tables 9 and 10. The bond distances and angles and the corresponding standard deviations were calculated using the full variance-covariance matrix of the least-squares refinement (Table 11). The numbering diagram for '*syn*'-*p*-chlorobenzaldoxime is shown in Fig. 1(b). The distances and angles obtained from the refined structure are, as would be expected, very close to the values reported earlier. The average C-C distance in the benzene ring is 1.384 Å and the average angle not involving H in the ring is 119.99°.

Table 8. Final values of $R = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|$
for 'syn'-*p*-chlorobenzaldoxime

<i>hkl</i>	<i>R</i>	Range of $\sin \theta$	<i>R</i>
All orders	0.118	0.00 to 0.40	0.112
even	0.124	0.40 to 0.50	0.106
odd	0.114	0.50 to 0.60	0.119
even	0.123	0.60 to 0.65	0.096
odd	0.115	0.65 to 0.70	0.109
even	0.124	0.70 to 0.75	0.114
odd	0.114	0.75 to 0.80	0.109
+ <i>l</i> even	0.120	0.80 to 0.85	0.109
+ <i>l</i> odd	0.117	0.85 to 0.90	0.119
+ <i>l</i> even	0.112	0.90 to 1.00	0.163
+ <i>l</i> odd	0.125		
+ <i>k</i> even	0.116		
+ <i>k</i> odd	0.121		
+ <i>k+l</i> even	0.122		
+ <i>k+l</i> odd	0.115		

Overall disagreement

$$R_F = 0.116$$

$$R_{F2} = 0.202$$

$$R_{wF2} = 0.298$$

The two sets of two-dimensional data for the 'anti'-*p*-chlorobenzaldoxime (Jerslev, 1957) were refined by least-squares with individual isotropic thermal parameters. As can be seen from the resulting parameters given in Table 12 the *z* coordinates

Table 9. Atomic parameters for 'syn'-*p*-chlorobenzaldoxime*

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$
Cl(1)	0.0921	-0.1305	0.2014	0.0002	0.0004	0.0001
C(2)	0.2524	0.1014	0.1671	0.0009	0.0012	0.0002
C(3)	0.4584	0.1679	0.1886	0.0010	0.0011	0.0002
C(4)	0.5900	0.3542	0.1612	0.0010	0.0013	0.0002
C(5)	0.5082	0.4681	0.1126	0.0009	0.0013	0.0002
C(6)	0.3042	0.3938	0.0909	0.0010	0.0014	0.0002
C(7)	0.1737	0.2076	0.1176	0.0010	0.0013	0.0002
C(8)	0.6544	0.6700	0.0861	0.0011	0.0015	0.0003
N(9)	0.5839	0.8095	0.0458	0.0009	0.0012	0.0002
O(10)	0.7505	0.9864	0.0276	0.0008	0.0013	0.0002

Scale factor: $s = 0.904 \pm 0.004$ (as previously defined)

* Unit-cell parameters are $a = 6.06$, $b = 4.73$, $c = 25.06$ Å and $\beta = 93.4^\circ$. The space group is $P2_1/c$, and there are four molecules in the unit cell (Jerslev, 1958).

Table 10. Anisotropic thermal parameters for 'syn'-*p*-chlorobenzaldoxime

The anisotropic thermal parameters are in the form:

$$\exp \{ - (h^2\beta_{11} + k^2\beta_{22} + l^2\beta_{33} + 2hk\beta_{12} + 2hl\beta_{13} + 2kl\beta_{23}) \}$$

Atom	β_{11}	β_{22}	β_{33}	β_{12}	β_{13}	β_{23}
Cl(1)	0.0306	0.0522	0.0017	-0.0094	0.0007	0.0009
C(2)	0.0272	0.0324	0.0016	0.0024	0.0004	-0.0004
C(3)	0.0302	0.0308	0.0016	-0.0034	-0.0005	0.0005
C(4)	0.0296	0.0484	0.0016	-0.0022	-0.0006	-0.0003
C(5)	0.0252	0.0467	0.0013	-0.0006	-0.0004	-0.0001
C(6)	0.0319	0.0507	0.0012	-0.0049	-0.0006	0.0008
C(7)	0.0290	0.0442	0.0016	-0.0020	-0.0004	-0.0002
C(8)	0.0313	0.0550	0.0017	-0.0054	-0.0002	0.0003
N(9)	0.0313	0.0508	0.0015	-0.0092	-0.0005	0.0005
O(10)	0.0364	0.0750	0.0019	-0.0213	-0.0005	0.0028

Table 11. Intramolecular bonds and angles for 'syn'-*p*-chlorobenzaldoxime

Atoms	Distance	Standard deviation*
Cl(1)-C(2)	1.728 Å	± 0.006 Å
C(2)-C(3)	1.367	0.008
C(3)-C(4)	1.396	0.008
C(4)-C(5)	1.396	0.008
C(5)-C(6)	1.367	0.008
C(6)-C(7)	1.382	0.008
C(7)-C(2)	1.396	0.008
C(5)-C(8)	1.486	0.009
C(8)-N(9)	1.260	0.008
N(9)-O(10)	1.408	0.007

Atoms	Angle	Standard deviation*
Cl(1)-C(2)-C(3)	118.5°	± 0.4
C(2)-C(3)-C(4)	119.1	0.5
C(3)-C(4)-C(5)	119.3	0.5
C(4)-C(5)-C(6)	121.0	0.5
C(5)-C(6)-C(7)	120.1	0.5
C(6)-C(7)-C(2)	119.0	0.5
C(7)-C(2)-C(3)	121.6	0.5
C(7)-C(2)-Cl(1)	119.9	0.4
C(4)-C(5)-C(8)	116.6	0.5
C(6)-C(5)-C(8)	122.5	0.5
C(5)-C(8)-N(9)	120.8	0.5
C(8)-N(9)-O(10)	110.8	0.4

Hydrogen bond distance
O-H ··· N 2.825 ± 0.006 Å

* The standard deviations are calculated using the full variance-covariance matrix from the least-squares refinement. The values should be multiplied by 2 or 3 for significant uncertainty.

Table 12. Atomic parameters for 'anti'-*p*-chlorobenzaldoxime*

0kl Refinement						
Atom	<i>y</i>	<i>z</i>	<i>B</i>	$\sigma(y)$	$\sigma(z)$	$\sigma(B)$
Cl(1)	0.3207	0.0528	4.89	0.0018	0.0004	0.23
C(2)	0.4466	0.0898	3.03	0.0049	0.0009	0.49
C(3)	0.6506	0.0627	3.37	0.0053	0.0009	0.59
C(4)	0.7493	0.0905	2.75	0.0047	0.0009	0.48
C(5)	0.6398	0.1457	2.43	0.0046	0.0007	0.46
C(6)	0.4286	0.1692	2.74	0.0052	0.0008	0.48
C(7)	0.3304	0.1434	3.29	0.0057	0.0009	0.60
C(8)	0.7594	0.1723	2.80	0.0046	0.0009	0.45
N(9)	0.6570	0.2111	3.32	0.0042	0.0009	0.44
O(10)	0.3948	0.2338	3.49	0.0037	0.0007	0.38

Scale factor $s = 0.895 \pm 0.030$						
<i>h0l</i> Refinement						
Atom	<i>x</i>	<i>z</i>	<i>B</i>	$\sigma(x)$	$\sigma(z)$	$\sigma(B)$
Cl(1)	0.0002	0.0527	5.24	0.0012	0.0003	0.23
C(2)	0.7782	0.0872	3.41	0.0036	0.0009	0.56
C(3)	0.6730	0.0630	6.61	0.0057	0.0013	0.91
C(4)	0.4924	0.0917	4.62	0.0045	0.0009	0.59
C(5)	0.4471	0.1436	2.35	0.0031	0.0007	0.39
C(6)	0.5555	0.1678	3.57	0.0041	0.0008	0.54
C(7)	0.7497	0.1427	6.00	0.0059	0.0011	0.80
C(8)	0.2412	0.1698	5.63	0.0048	0.0010	0.71
N(9)	0.1432	0.2130	3.88	0.0034	0.0010	0.52
O(10)	0.2027	0.2357	5.73	0.0034	0.0009	0.61

Scale factor $s = 0.930 \pm 0.040$						
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* Unit-cell parameters for 'anti'-*p*-chlorobenzaldoxime are $a = 6.60$, $b = 4.67$ and $c = 23.52$ Å. The space group is $P2_12_12_1$, and there are four molecules in the unit cell (Jerslev, 1958).

Table 13. Disagreement indices for '*anti*'-*p*-chlorobenzaldoxime

0kl Refinement			
Class	R_F	$\sin \theta$	R_F
All orders	0.134	0.00 to 0.40	0.176
<i>k</i> even	0.142	0.40 to 0.50	0.114
<i>k</i> odd	0.129	0.50 to 0.60	0.101
<i>l</i> even	0.129	0.60 to 0.65	0.151
<i>l</i> odd	0.139	0.65 to 0.70	0.130
<i>k+l</i> even	0.140	0.70 to 0.75	0.121
<i>k+l</i> odd	0.129	0.75 to 0.80 0.80 to 0.85 0.85 to 0.90 0.90 to 1.00	0.081 0.193 0.109 0.166
Overall disagreement			
	$R_F = 0.145$	$R_{F2} = 0.28$	$R_{wF2} = 0.33$
<i>h0l</i> Refinement			
Class	R_F	$\sin \theta$	R_F
All orders	0.197	0.00 to 0.40	0.145
<i>h</i> even	0.201	0.40 to 0.50	0.172
<i>h</i> odd	0.195	0.50 to 0.60	0.187
<i>l</i> even	0.207	0.60 to 0.65	0.237
<i>l</i> odd	0.188	0.65 to 0.70	0.260
<i>h+l</i> even	0.160	0.70 to 0.75	0.230
<i>h+l</i> odd	0.254	0.75 to 0.80 0.80 to 0.85 0.85 to 0.90 0.90 to 1.00	0.240 0.235 0.236 0.242
Overall disagreement			
	$R_F = 0.182$	$R_{F2} = 0.26$	$R_{wF2} = 0.42$

suggest that, although the basic structure is correct, the revised bond distances are not of sufficient accuracy to justify comparison with the other two structures. Accordingly, bond distances and angles will not be reported until a revised structure is obtained from three-dimensional data. The final disagreement indices R for the two sets of data are given, however, in Table 13. For the *h0l* data $R = 0.182$, $R_{F2} = 0.26$, $R_{wF2} = 0.42$ and for the *0kl* data $R = 0.144$, $R_{F2} = 0.28$ and $R_{wF2} = 0.33$.

Comparison of the *N*-methyl oxime with the '*syn*' and '*anti*' oximes

The Cl-C distances of 1.768 Å in the *N*-methyl oxime and of 1.728 Å in the '*syn*' oxime differ by 0.04 Å, which is some six times the standard deviations based upon the assumption of random errors. Nevertheless, there may be some systematic problems near the relatively heavy Cl atoms and hence we are reluctant to press a claim for this difference, which we do not understand. Also, the C-C bond distances in the benzene rings vary from extremes of 1.363 Å to 1.411 Å in the *N*-methyl oxime, and from 1.367 Å to 1.396 Å in the '*syn*' oxime, and hence we have

suggested a factor of 2 or 3 times standard deviations as a requirement for significant uncertainty. In the oxime part of the molecule the N → O distance of 1.284 Å in the *N*-methyl oxime and the N-O distance of 1.408 Å in the '*syn*' oxime are, of course, quite significantly different in the expected direction. Moreover, the C=N distance of 1.309 Å in the *N*-methyl oxime is 0.05 Å longer than the C=N distance of 1.260 Å in the '*syn*' oxime, and we believe that this difference is significant partly because it is far removed from the Cl atom, and partly because it seems understandable in view of the reverse trend of the N-O distances in these two compounds.

Comparison of the bond angles in these two compounds is also interesting. In the *N*-methyl oxime, $\angle C(6)C(5)C(8) = 127.8^\circ$, $\angle C(5)C(8)N(9) = 124.7^\circ$ and $\angle C(8)N(9)O(10) = 125.2^\circ$ are all increased above 120° in such a way that the H atom attached to C(6) is 2.2 Å away from O(10). Even this H ··· O distance is about 0.4 Å less than the van der Waals contact of 2.6 Å which is found from intermolecular, rather than intramolecular, contacts. It is clear that if the CH₃ and O are reversed by rotation about the C=N bond, there are then H ··· H contacts between the H atom attached to C(6) and two H atoms of the CH₃ group of only 1.6 Å, which is 0.8 Å less than the usual van der Waals contact. Certainly, then, it is no surprise that a planar '*anti*' configuration is more stable than the planar '*syn*' configuration for the *N*-methyl oxime; but, since in the non-methylated '*anti*' oxime a plane through C(8), N(9) and O(10) deviates considerably (19°) (Jerslev, 1958) from the plane of the benzene ring, it is surprising that the *N*-methylated '*anti*' form is so nearly coplanar, except, of course, for some H atoms of the CH₃ group. In view of these short contacts, it is probable that if the as yet unknown '*syn*' form of the *N*-methyl oxime were discovered it would not have the C, N and O atoms in coplanar arrangement. The external C-C bond is displaced from the C(2)C(5) bisector of the benzene ring in both the '*syn*' and '*anti*' oximes in the direction expected from steric considerations, and, also as expected, this displacement is comparatively smaller in the '*syn*' oxime. The C(6)C(5)C(8) angle in the '*syn*' oxime is only 122.5° (only 2.5° above 120°), and the distance from N(9) to the H atom attached to C(6) is a reasonable 2.6 Å.

In Table 14 we summarize bond distances, which are in common among a number of oximes and hydrochlorides of amine oxides. In trimethylamine oxide the N → O distances of 1.36 ± 0.03 Å (Lister & Sutton, 1939) and 1.44 ± 0.04 Å (Rundle, 1950) have been reported. In the present study, where this N → O bond is much more highly conjugated it is remarkably short, much shorter in the *N*-methyl oxime than the N-O distances found in hydrochlorides of amine oxides.

Table 14. Comparison of distances in some oximes and amine oxides

Compound	N-O	C=N	Reference
Acetoxime	1.36 Å	1.29 Å	Bierlein & Lingafelter (1951)
Dimethylglyoxime*	1.32	1.25	Hamilton (1961)
5-Chlorosalicylaldoxime	1.385	1.237	Simonsen, Pfluger & Thompson (1961)
'Syn'- <i>p</i> -chlorobenzaldoxime	1.408	1.260	This study
'Anti'- <i>p</i> -chlorobenzaldoxime	1.39	1.26	Jerslev (1957)
<i>N</i> -Methyl- <i>p</i> -chlorobenzaldoxime	1.284	1.309	This study
Trimethylamine oxide†	{ 1.36 1.44	— }	Allen & Sutton (1950)
Trimethylamine oxide HCl	1.425	—	Caron & Donohue (1962)
4-Nitropyridine- <i>N</i> -oxide	1.260	—	Eichhorn (1956)
4,4'- <i>Trans</i> -azopyridine- <i>N</i> -oxide	1.283	—	Eichhorn (1959)
Pyridinoxide HCl	1.375	—	Tsoucaris (1961)
Formaldoxime‡	1.408	1.276	Levine (1963)
Trimethylamine oxide†	{ 1.36 1.44	—	{ Lister & Sutton (1939) Rundle (1950)

* Neutron diffraction study.

† Electron diffraction studies.

‡ Microwave study.

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